



NEET 2023-24

Mark
720

Group
PCB

PRE FINAL ROUND -01

Date : 17/03/2024

Time : 3 :20 Hours

Answer Key Version - S (PCB NEET 2023-24)

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PHYSICS

SECTION - A (35 Questions)

01. (4) $a = \frac{S^2}{t^4} = \frac{(\text{metre})^2}{(\text{second})^4} = \text{m}^2 \text{s}^{-4}$

02. (1) $\text{KE} = \frac{1}{2} mV^2$

$[\text{KE}] = \text{ML}^2 \text{T}^{-2}$

If unit of M and L are doubled

Then unit of K.E.

$\text{K.E.} = [(2M) (2L)^2 \text{T}^{-2}]$

$= 8 [\text{ML}^2 \text{T}^{-2}]$

unit of K.E. is 8 times.

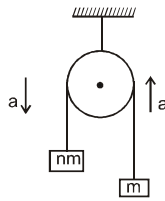
03. (3) $a = \frac{(nm - m)}{nm + m} g$

$= \frac{(n-1)}{(n+1)} g$

$a_1 = a_2 = a$

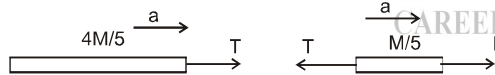
$a_{cm} = \frac{nma_1 - ma_2}{(nm + m)} = \frac{(n-1)}{(n+1)} \times a$

$a_{cm} = \frac{(n-1)^2}{(n+1)^2} g$



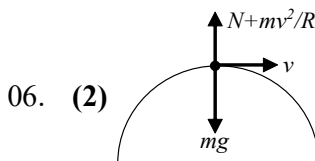
04. (1) Velocity can't change its value suddenly.

05. (3) $a = \frac{F}{M}$



By FBD

$T = \frac{4M}{5} \times a = \frac{4M}{5} \times \frac{F}{M} = \frac{4F}{5} \Rightarrow T = 4 \text{ N}$



06. (2)

For leaving contact $N = 0$

$\Rightarrow \frac{mv^2}{R} = mg \Rightarrow v = \sqrt{gR}$

07. (3) use $m_1 v_1 = m_2 v_2 = P$

$\text{K.E.} = \frac{1}{2} m_1 v_1^2 + \frac{1}{2} m_2 v_2^2$

$= \frac{1}{2} m_1 \left(\frac{P}{m_1}\right)^2 + \frac{1}{2} m_2 \left(\frac{P}{m_2}\right)^2$

$= \frac{1}{2} \frac{P^2(m_2 + m_1)}{m_1 m_2}$

08. (2) $x = 6t$

$y = 8t - 5t^2$

$\frac{dx}{dt} = 6$

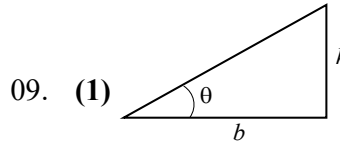
$\frac{dy}{dt} = 8 - 10t$

at $t = 0$

$V_x = 6 \text{ m/sec}$

$V_y = 8 \text{ m/sec}$

$V = \sqrt{V_y^2 + V_x^2} = \sqrt{8^2 + 6^2} = 10 \text{ m/sec}$

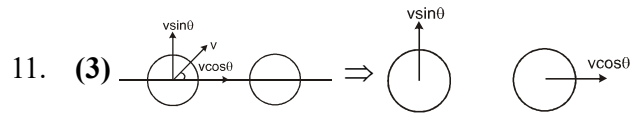


09. (1)

as $v = \sqrt{Rg \tan \theta} \Rightarrow h = \frac{v^2 b}{Rg}$

10. (2) $\frac{H}{R} = \frac{\tan \theta}{4}$

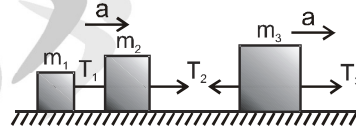
$\theta = 45^\circ$ & $R = 36 \text{ m} \therefore H = 9 \text{ m}$.



11. (3)

before collision after collision
So angle between velocity vectors is 90°

12. (3) $a = \frac{60}{10 + 20 + 30} = 1 \text{ ms}^{-2}$



$\Rightarrow T_2 = (m_1 + m_2) a = (10 + 20) \times 1 = 30 \text{ N}$

13. (4) It can be observed that component of acceleration perpendicular to velocity is $a_c = 5 \text{ m/s}^2$

$\therefore \text{radius} = \frac{v^2}{a_c} = \frac{25}{5} = 5 \text{ metre}$

14. (4)

15. (3) Let v be the speed of B at lowermost position, the speed of A at lowermost position is $2v$.

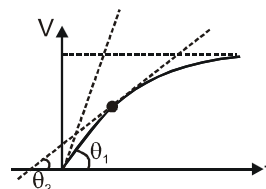
From conservation of energy

$\frac{1}{2} m (2v)^2 + \frac{1}{2} m v^2 = mg(2l) + mgl$

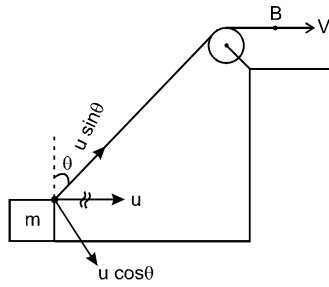
Solving we get $v = \sqrt{\frac{6}{5} gl}$

16. (1) As the slope of tangent decreases, velocity also decreases with time.

after time distance becomes constant i.e particle stops.



17. (2)



The length of string AB is constant.

⇒ speed A and B along the string are same $u \sin \theta = V$

$$u \sin \theta = V$$

$$u = \frac{V}{\sin \theta}$$

18. (3) $[Y] = [F^a A^b D^c]$

$$[ML^{-1}T^{-2}] = [(MLT^{-2})^a (L^2)^b (ML^{-3})^c]$$

equating power of M, L and T

$$1 = a + c, \quad -1 = a + 2b - 3c$$

$$-2 = -2a \quad a = 1, \quad c = 0$$

$$b = -1$$

$$[Y] = F A^{-1} D^0.$$

19. (2) $[h] = ML^2T^{-1}$

$$[V_s] = \frac{[W]}{[Q]} = \frac{ML^2T^{-2}}{AT} = ML^2T^{-3} A$$

$$[\phi] = ML^2T^{-2}$$

$$[P] = MLT^{-1}.$$

20. (2)

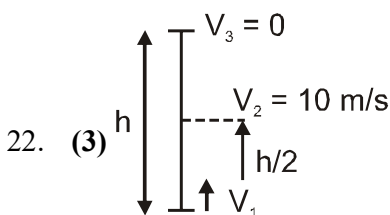
21. (3) $W_G - W_f = 0 \Rightarrow mgh = \mu mgl$

$$h = \mu l$$

$$h = (0.2)l$$

$$\Rightarrow l = \frac{1.5}{0.2}$$

$$l = 7.5 \text{ m} = (3 + 3 + 1.5) \text{ m}.$$



22. (3)

$$V_2^2 = V_3^2 + 2g \frac{h}{2}$$

$$\Rightarrow (10)^2 = 2 \times 10 \times \frac{h}{2}$$

$$\Rightarrow h = \frac{100}{10} = 10 \text{ m}$$

23. (3) It can be observed that power delivered to particle by force F is -

$$P = Fv = K.$$

The power is constant. Hence work done by force in time t is -

$$\Delta W = Pt = Kt$$

24. (1) $W_s + W_f = \Delta K$

$$-\Delta U + W_f = -K_i$$

$$-U_f - \mu mgx = -K_i$$

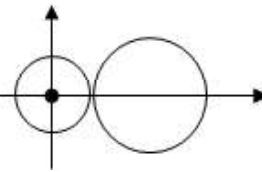
$$\frac{1}{2} Kx^2 + \mu mgx = \frac{1}{2} mu^2$$

$$100x^2 + 2(0.1)(50)(10)x = 50 \times 4$$

$$x^2 + x - 2 = 0$$

$$x = 1 \text{ m}.$$

25. (2)



$$X_{com} = \frac{m \times 0 + 2m \times (3a)}{m + 2m} = 2a.$$

26. (2) $a = bt$

$$\frac{dv}{dt} = bt$$

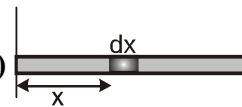
$$\int_u^v dv = \int_0^t bt dt$$

$$V = u + \frac{bt^2}{2}$$

$$[V = u + \frac{bt^2}{2}]$$

27. (1) $W = \vec{F} \cdot \vec{s} = (-2\hat{i} + 15\hat{j} + 6\hat{k}) \cdot 10\hat{j} = 150 \text{ J}.$

28. (2)

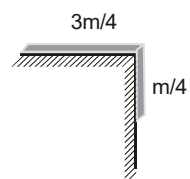


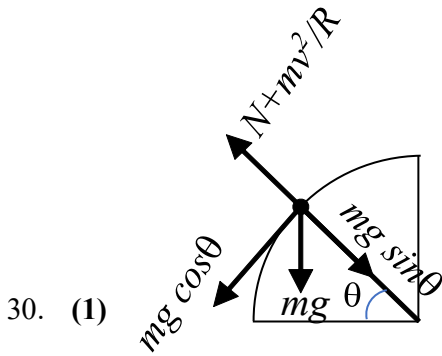
$$x_{cm} = \frac{\int dm x}{\int dm} = \frac{\int (\lambda_0 x dx) x}{\int \lambda dx} = \frac{\int_0^L \lambda_0 x^2 dx}{\int_0^L \lambda_0 x dx} = \frac{2L}{3}.$$

29. (1) Apply system equation

$$\frac{m}{4} g = \frac{3m}{4} g \times \mu$$

$$\Rightarrow \mu = \frac{1}{3} = 0.33$$

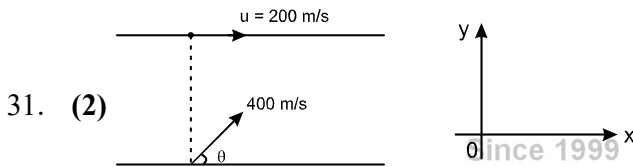




$$N + \frac{mv^2}{R} = mg \sin \theta$$

$$N = mg \sin \theta - \frac{mv^2}{R}$$

as $\theta \uparrow \Rightarrow N \uparrow$



To hit, $400 \cos \theta = 200$

{ \because Both travel equal distance along horizontal, of their start and coordinates on x axis are same}

$$\Rightarrow \theta = 60^\circ$$

32. (3)
$$V_{com} = \frac{200 \times 10 \hat{i} + 500(3 \hat{i} + 5 \hat{j})}{200 + 500}$$

$$= \frac{20 \hat{i} + 15 \hat{i} + 25 \hat{j}}{7} = 5 \hat{i} + \frac{25}{7} \hat{j}$$

33. (1) $V = \sqrt{5lg} = \sqrt{5 \times 6.4 \times 10} = 17.9 \text{ m/s}$

34. (4) $\Delta V = V_2 - V_1$
 $= 0$

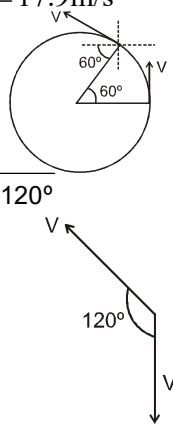
$$\Delta \vec{V} = \vec{V}_2 - \vec{V}_1$$

$$|\Delta \vec{V}| = \sqrt{V_2^2 + V_1^2 + 2V_1V_2 \cos 120^\circ}$$

$$= \sqrt{V^2 + V^2 + 2V^2 \cos 120^\circ}$$

$$= \sqrt{2V^2 + 2V^2 \left(\frac{-1}{2}\right)}$$

$$|\Delta \vec{V}| = V$$



35. (2)
$$X_{CM} = \frac{0 \times m + m \times a + m \times \frac{a}{2}}{m + m + m} = \frac{a}{2}$$

$$Y_{CM} = \frac{0 \times m + 0 \times m + m \times \frac{a\sqrt{3}}{2}}{m + m + m} = \frac{a\sqrt{3}}{6}$$

SECTION - B (Attempt Any 10 Questions)

36. (1) As block is shifted slowly $\Delta K.E. = 0$

$$\therefore W_g + W_f + W_F = 0$$

Work done :

$$= Mgh_1 + Mgh_2 + Mgh_3 + \mu_1 Mgl_1 + \mu_2 Mgl_2 + \mu_3 Mgl_3$$

$$= Mg(h_1 + h_2 + h_3) + Mg(\mu_1 l_1 + \mu_2 l_2 + \mu_3 l_3)$$

$$= Mg(8 + 0.2 + 0.4 + 0.4) = 90 \text{ J}$$

37. (2) $W = \int \vec{F} \cdot d\vec{s} = \int (3t\hat{i} + 5\hat{j}) \cdot (4t dt \hat{i})$

$$= \int_0^2 12t^2 dt = \frac{12[t^3]_0^2}{3} = 32 \text{ J}$$

38. (1) $m\vec{V}_m = -M\vec{V}_b$

$$m(\vec{V}_{rel} + \vec{V}_b) = -M\vec{V}_b$$

$$\vec{V}_b = \frac{-m\vec{V}_{rel}}{M + m}$$

$\Rightarrow \vec{V}_b$ will be opposite to V_{rel} .

39. (1) $\Delta U = \frac{1}{2} \frac{m_1 m_2}{(m_1 + m_2)} (V_1 - V_2)^2 = \frac{100}{3}$

$$(V_1 - V_2)^2 \times \frac{2m \cdot m}{2(m + 2m)} = \frac{100}{3}$$

putting $m = 1 \text{ kg}$

$$(V_1 - V_2) = 10 \text{ m/sec}$$

40. (2) Case (1) :

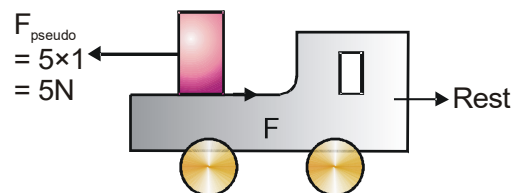
$$a = \frac{F}{3m}$$

$$N_1 = m \times a$$

Similarly in case (2)

$$N_2 = 2m \times a \Rightarrow \frac{N_1}{N_2} = \frac{1}{2}$$

41. (1) Solving from the frame of truck



$$f \leq \mu mg = 6$$

$$\Rightarrow f = 5 \text{ N}$$

42. (4) $0.2 \times 100 \text{ g} \leftarrow$ $0.3 \times 300 \text{ g} \leftarrow$ \boxed{B} $\rightarrow F$
for motion to start
 $F \geq 0.2 \times 100 \text{ g} + 0.3 \times 300 \text{ g} = 1100 \text{ N}$
 $F_{\min} = 1100 \text{ N}$

43. (1) $N = mg + Q \cos \theta$
frictional force $f = \mu(mg + Q \cos \theta)$
 $P + Q \sin \theta = \mu(mg + Q \cos \theta)$

$$\mu = \frac{P + Q \sin \theta}{mg + Q \cos \theta}$$

44. (1) From given conditions :

$$V_A = V_B \cos 37^\circ = 15 \cdot \frac{4}{5} = 12 \text{ m/sec.}$$

$$\therefore \text{time of flight of A (t)} = \sqrt{\frac{2 \times 20}{10}} = 2 \text{ sec.}$$

$$\Rightarrow \text{Range} = V_A t = 24 \text{ m.}$$

45. (3)

46. (3) Density, $\rho = \frac{m}{V}$

$$\Rightarrow \left| \frac{\Delta \rho}{\rho} \right|_{\max} = \frac{m}{\pi r^2 l} = \left| \frac{\Delta m}{m} \right| + 2 \left| \frac{\Delta r}{r} \right| + \left| \frac{\Delta l}{l} \right|$$

$$= \frac{0.01}{0.4} + \frac{2(0.03)}{6} + \frac{0.04}{8}$$

$$\% \text{ error in density} = \left(\frac{\Delta \rho}{\rho} \right) \times 100\%$$

$$= \left(\frac{1}{0.4} + \frac{6}{6} + \frac{4}{8} \right) \% = (2.5 + 1 + 0.5) \% = 4\%$$

47. (3) MSR = 2.5 mm

$$CSR = 45 \times \frac{0.5}{50} \text{ mm} = 0.45 \text{ mm}$$

$$\text{Diameter reading} = \text{Reading of crew gauge} \\ = 2.5 + 0.45 - (-0.03) = 2.98 \text{ mm.}$$

48. (1) $\frac{dx}{dt} = \text{slope} \geq 0$ always increasing

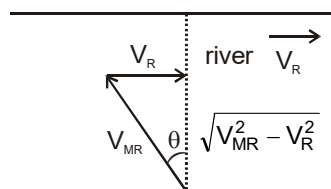
$$\frac{dx}{dt} < 0 ; \text{ and at } t \rightarrow \infty \frac{dx}{dt} \rightarrow 0$$

$$\frac{dx}{dt} > 0 \text{ for first half } \frac{dx}{dt} < 0 \text{ for second half.}$$

$$\frac{dx}{dt} = \text{constant}$$

49. (1) Work done by a force is positive if displacement is in direction of force and work done by a force is negative if displacement is in direction opposite to that of force.

50. (2) 15 min = 1/4 hr.



$$t = \frac{d}{V_y}$$

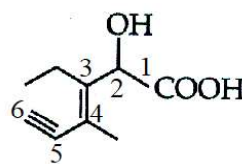
$$\Rightarrow \frac{1}{4} = \frac{1}{\sqrt{V_{MR}^2 - V_R^2}} = \frac{1}{4} = \frac{1}{\sqrt{5^2 - V_R^2}}$$

$$\Rightarrow V_R = 3 \text{ km/h}$$

CHEMISTRY

SECTION - A (35 Questions)

51. (4) Number of moles of oxygen = 2 × number of moles of given compounds.
52. (1) 1
53. (4) If both assertion and reason are false.
54. (4) E⁺ attacks on ring which has more e⁻ density.
55. (2) 1 is staggered and 2 is eclipsed.
56. (2) 2s
57. (2) As values of m is from -1 to +1 including zero.
58. (4) All the above
59. (2) Higher are number of α-H, more the hyperconjugating structures, more the stability of the compound.
60. (1) 1, 2 and 3
61. (4) Helium nuclei, which impinged on a metal foil and got scattered.
62. (1) 1
63. (4)



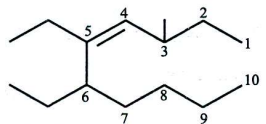
64. (4) (A) is elimination, (B) is substitution and (C) is addition reaction.
65. (4) (1)-(iv), (2)-(iii), (3)-(ii), (4)-(i)
66. (4)

$$X_3 = \frac{X_1 X_2}{X_1 + X_2}$$

67. (3) (1)-(iv), (2)-(ii), (3)-(i), (4)-(iii)
68. (3) Four primary amines are possible. These are: CH₃CH₂CH₂CH₂NH₂, (CH₃)₂CH-CH₂NH₂, CH₃CH(NH₂)CH₂CH₃ and (CH₃)₃CNH₂.
69. (1) 9σ and 9π

70. (4) Statement-I is incorrect and Statement-II is correct
71. (4) (1)-(iv), (2)-(iii), (3)-(ii), (4)-(i)
72. (2) 3, 3 and 3 respectively
73. (2) $-\text{COOH}$, $-\text{SO}_3\text{H}$, $-\text{CONH}_2$, $-\text{CHO}$
74. (3) (A) \rightarrow (iii), (B) \rightarrow (iv), (C) \rightarrow (ii), (D) \rightarrow (i)
75. (3) Three, that is, $\text{CH}_3\text{OCH}_2\text{CH}_2\text{CH}_3$, $\text{CH}_3\text{O}-\text{CH}(\text{CH}_3)_2$ and $\text{CH}_3\text{CH}_2\text{OCH}_2\text{CH}_3$.
76. (3)
Wt. of solvent = Wt. of solution – Wt. of solute
= $[1000 \times 1.02 - 20.5 \times 60] = 897 \text{ g}$.
$$m = \frac{\text{Moles of } \text{CH}_3\text{COOH}}{\text{Wt. of solvent in kg}} = \frac{2.05 \times 1000}{897}$$

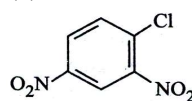
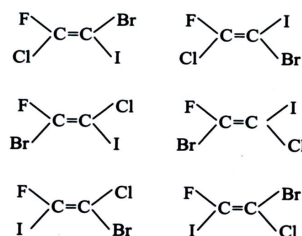
= 2.285
77. (2)
Mol. wt. of H_3PO_4 is 98 and change in its valence = 1.
$$\text{Eq. wt. of } \text{H}_3\text{PO}_4 = \frac{\text{Mol. wt.}}{\text{Change in valency}}$$

= $98/1 = 98$
78. (2)

5, 6-Diethyl-3-methyldec-4-ene
79. (3)
 NO_2^+ , AlCl_3 , SO_3 and $\text{CH}_3\overset{+}{\text{C}}=\text{O}$ are electrophiles.
80. (1)
Three, that is, d-tartaric acid, l-tartaric acid and meso-tartaric acid.
81. (1)
 $2\text{NaHCO}_3 \longrightarrow \text{Na}_2\text{CO}_3 + \text{H}_2\text{O} + \text{CO}_2$
2 mol of NaHCO_3 on complete decomposition gives 1 mol of Na_2CO_3 .
So, 0.2 mol of NaHCO_3 on complete decomposition gives 0.1 mol of Na_2CO_3 .
82. (4)
According to stoichiometry, they should react as follows:
 $4\text{NH}_3(\text{g}) + 5\text{O}_2(\text{g}) \rightarrow 4\text{NO}(\text{g}) + 6\text{H}_2\text{O}(\text{l})$
4 mol 5 mol 4 mol 6 mol
0.8 mol 1 mol 0.8 mol 1.2 mol
In this reaction 1 mole of O_2 and 0.8 mole of NH_3 are consumed. There by indicating complete consumption of O_2 .
83. (3)
Order of stability of carbanions is $1^\circ > 2^\circ > 3^\circ$.
84. (1) II > I > III
85. (3) As both the carbon atoms of each of the three double bonds are differently substituted, therefore, $2^3 = 8$ geometrical isomers are possible.

SECTION - B (Attempt Any 10 Questions)

86. (3) Statement-I is correct but Statement - II is incorrect. Zeros at the end or right of a number are significant provided they are on the right side of the decimal point.
87. (1) Molecular weight of the metal chloride
$$= \frac{0.72 \times 22400}{100} = 161.28 \text{ g}$$

Weight of chlorine in metal chloride
$$= \frac{65.5 \times 161.28}{100} = 105.64 \text{ g}$$

So, Mole atoms of chlorine = $\frac{105.64}{35.5} = 3$
Hence, metal chloride is MCl_3
88. (4) 4-chloro-3-ethylcyclohexanol
89. (3)

4-chloro-1,3-dinitrobenzene
90. (3) Charge of electron
91. (1) As halogens are most electronegative so the configuration is $ns^2 np^5$.
92. (2)
Carbanions are stabilised by electron withdrawing groups. $-\text{NO}_2$ is stronger electron withdrawing group as compared to $-\text{CHO}$. At ortho-position, the effect is more pronounced.
93. (3) $-\text{NO}_2$ group is meta-directing, thus will stabilize a electrophile at m-position.
94. (3)
$$\lambda = \frac{h}{\sqrt{2m(\text{KE})}} = 0.3328 \text{ nm}$$
95. (3) As maximum number of electrons in any orbit, sub-orbit or orbital is decided by Pauli's law.
96. (1) Non-superimposable on its mirror image.
97. (4) The two stereoisomers are not mirror images and hence, the diastereomers.
98. (4) Six isomers are

99. (3) 2 and 3
100. (3)
