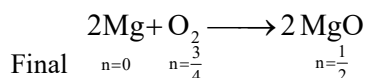
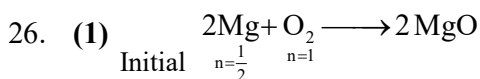


R ANSWER KEY & SOLUTION KEY FINAL ROUND - 07 (PCB) Dt.15.04.2024

CHEMISTRY

SECTION - A (35 Questions)

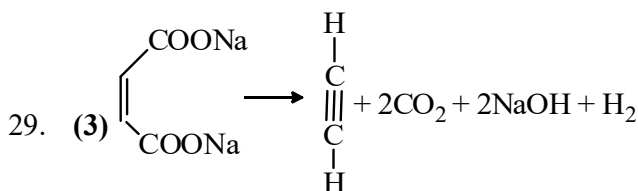
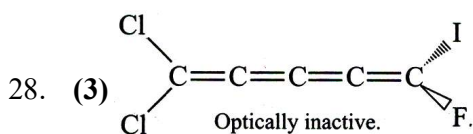
01. (1) $2\text{CrO}_4^{2-} + 2\text{H}^+ \rightarrow \text{Cr}_2\text{O}_7^{2-} + \text{H}_2\text{O}$
 $\text{Cr}_2\text{O}_7^{2-} + 2\text{OH}^- \rightarrow 2\text{CrO}_4^{2-} + \text{H}_2\text{O}$
02. (1) 0.85 Å
03. (4) The azeotropic mixture cannot be separated into individual components as both the components boil at the same temperature.
04. (2) At equilibrium, rate of forward reaction is equal to the rate of backward reaction. As the equilibrium is dynamic, reaction will not stop ($r_f \neq 0$ and $r_b \neq 0$)
05. (2) Cis addition occurs and molecule becomes Achiral
06. (4) A-iii, B-i, C-ii, D-iv
07. (3) As the size of anion decreases, the polarisability decreases (small cation, large anion, high charge, more covalent or more polarisability) (Fajans' rule).
08. (1) Apply Fajan's rule, charge on the cation is same and the anion is common in all the compounds, so the size of cation will decide the ionic character or the ionic character down group 2, increases. So, the order is as given in (1).
09. (1) (1) $[\text{Ag}^+] = \frac{K_{sp}}{[\text{Br}^-]} = 5 \times 10^{-12} \text{ M}$
 (2) $[\text{Ag}^+] = \frac{K_{sp}}{[\text{Cl}^-]} = 1.8 \times 10^{-7} \text{ M}$
 (3) $[\text{Ag}^+] = \sqrt{\frac{K_{sp}}{[\text{CO}_3^{2-}]}} = 9 \times 10^{-6} \text{ M}$
 (4) $[\text{Ag}^+] = \left(\frac{K_{sp}}{[\text{ASO}_4^{3-}]} \right)^{1/3} = 10^{-7} \text{ M}$
10. (2) In adiabatic process, there is no heat transfer between system and surrounding.
11. (4) If both assertion and reason are false.
12. (3) Electron deficient group decreases the electron density of N-atom, thus, makes its lone pair less available for donation.
13. (3) Triamminebromochloronitroplatinum (IV) chloride
14. (1) If both assertion and reason are true and reason is the correct explanation of the assertion.
15. (3) An extensive property is a property whose value depends on the quantity or size of matter present in the system. For example, mass, volume, internal energy, enthalpy, heat capacity, etc. are extensive properties.
16. (4) CaCO_3 is a salt with n-factor = 2
17. (3) In Hofmann elimination reaction, less substituted alkene is the main product.
- $$\left[\text{CH}_3\text{CH}_2\text{CH}_2-\overset{\text{CH}_3}{\underset{\text{CH}_3}{\text{N}^+}}-\text{CH}_2\text{CH}_3 \right] \text{OH}^- \xrightarrow{\Delta}$$
- $\text{CH}_3\text{CH}_2\text{CH}_2-\text{N}(\text{CH}_3)_2 + \text{CH}_3\text{CH}=\text{CH}_2 + \text{CH}_2=\text{CH}_2$
 (minor) (major)
18. (1) The genetic information of cell is contained in the sequence of base A, T, G and C in DNA molecule. When a cell divides, DNA molecules replicate and make exact copies of themselves so that each daughter cell will have DNA identical to that of the parent cell.
19. (4) Bohr's atomic theory is applicable to hydrogen and single electron species like He^+ , Li^{2+} , Be^{3+} , etc.
20. (4) For g orbital, maximum value of l is 4.
 Since $l = n - 1$, n should be 5.
21. (1) $D > B > A > C$
22. (3) Statement-1 is false, Statement-2 is true
23. (4) This is an intramolecular Cannizzaro reaction. On acidification, it gives a product containing an alcohols and a free acid group.
24. (3) Phthalic acid
25. (3) (a)-(ii); (b)-(iv); (c)-(iii); (d)-(i)



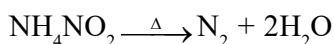
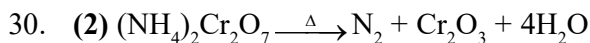
27. (3) Earlier rate = $kA^n B^m$

New rate = $k(2A)^n \left(\frac{1}{2}B\right)^m$

$\frac{\text{New rate}}{\text{Earlier rate}} = \frac{2^n A^n B^m 2^{-m}}{A^n B^m} = 2^n \cdot 2^{-m} = 2^{n-m}$



Acetylene + CO_2 is formed at anode

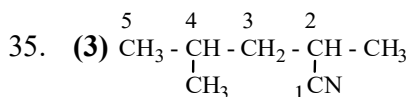


31. (2) Because yellow phosphorus is most reactive form phosphorus and red P is highly polymerised.

32. (2) Minus signs are for reactants and positive signs for products. Dividing numbers are the coefficients.

33. (4) Molar conductivity of an electrolyte $A_m B_n$ is $m\lambda_{(A^{n+})}^\infty + n\lambda_{(B^{m-})}^\infty$

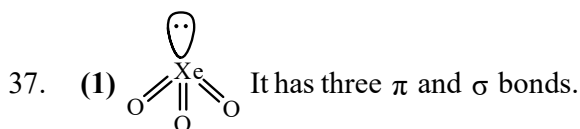
34. (3) The conversion of $-\text{CN}$ to $-\text{CH}_2\text{NH}_2$ by catalytic reduction is called Mendius reaction.



2, 4-dimethyl pentane nitrile

SECTION - B (Attempt Any 10 Questions)

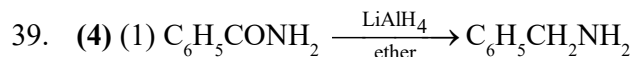
36. (1) (i)-(c), (ii)-(d), (iii)-(b), (iv)-(a)



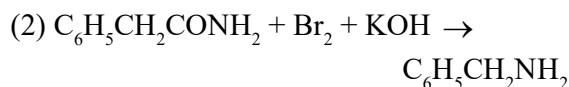
38. (2) In first reaction, O_2 is LR
In second reaction, N_2 is LR

In third reaction, Cl_2 is LR

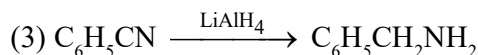
In fourth reaction, C is LR



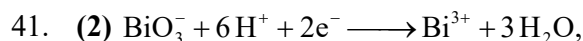
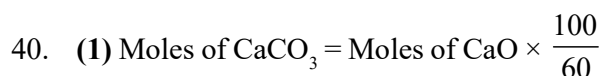
Benzylamine



Benzylamine



Benzylamine



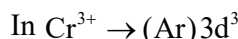
42. (3) 2

43. (3) The correct statement is : Hydrogen bond is stronger than the van der Waals forces.

44. (3) $\mu = \sqrt{n(n+2)}$

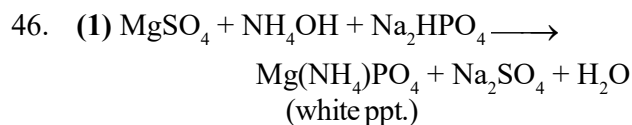
$3.87 = \sqrt{n(n+2)}$

$n = 3$



$n = 3$

45. (3) Statement-1 is false, Statement-2 is true

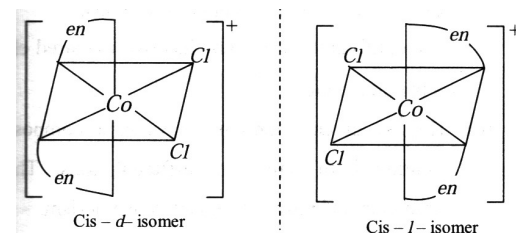


$\text{Rate} = -\frac{1}{2} \frac{d[\text{N}_2\text{O}_4]}{dt} = +\frac{1}{4} \frac{d[\text{NO}_2]}{dt}$

48. (1) The treatment with FeCl_3 yield ferric ferrocyanide which has bluish green or prussian blue colouration



49. (3) $[\text{Co}(\text{en})_2\text{Cl}_2]^+$

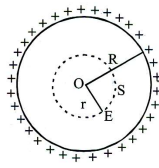


$$\int_s \vec{E} \cdot d\vec{s} = \frac{1}{\epsilon_0} \times \text{charge enclosed or}$$

$$E \times 4\pi r^2 = \frac{1}{\epsilon_0} \times 0$$

$$\Rightarrow E = 0$$

i.e. electric field inside a hollow sphere is zero.



63. (1) $B_1 \propto \frac{3I}{r}$ (Field due to AB, BC and CD)

$$B_2 \propto 3 \times \frac{I}{r} \text{ (Field due to AD)}$$

Clearly, the two fields are equal in magnitude and the opposite in direction.

64. (1) Given that $K_p = K_d = K_\alpha = K$ (say). We know that $m_p = m$, $m_d = 2m$ and $m_\alpha = 4m$ and $q_p = e$, $q_d = e$ and $q_\alpha = 2e$

$$\text{Further, } r = \frac{\sqrt{2mK}}{qB} \Rightarrow r_p = \frac{\sqrt{2mK}}{eB}$$

$$r_d = \frac{\sqrt{2(2m)K}}{eB} = \sqrt{2}r_p$$

$$\text{and } r_\alpha = \frac{\sqrt{2(4m)K}}{(2e)B} = r_p. \text{ Hence, } r_\alpha = r_p < r_d$$

65. (4) The situation is equivalent to a conductor of effective length $L = 2r$ falling in horizontal magnetic field. Therefore, induced emf $e = \text{pot. diff. developed} = BLv = B(2r)v = 2rBv$. The direction of current is from R to P . Therefore, R is at higher potential.

66. (1) $\frac{\epsilon_1}{\epsilon_2} = \frac{N_1}{N_2} = \frac{100}{10} \Rightarrow \epsilon_2 = 22 \text{ V}$

$$I = \frac{22}{44 \times 10^3} = \frac{1}{2} \text{ mA}, V_0 = \frac{1}{2} \text{ mA} \times 14 \text{ k}\Omega = 7 \text{ V}$$

67. (1) Stopping potential is a measure of maximum kinetic energy of emitted photoelectron ($eV_n = K_{\max}$) and K_{\max} depends upon the frequency of incident light but is independent of intensity.

68. (1) Energy of a H-like atom in its n th state is given by

$$E_n = -Z^2 \times \frac{13.6}{n^2} \text{ eV}$$

For, first excited state of He^+ , $n = 2$, $Z = 2$

$$\therefore E_{\text{He}^+} = -\frac{4}{2^2} \times 13.6 = -13.6 \text{ eV.}$$

69. (3)

70. (4) In first case, $W_1 = \frac{1}{2} kx^2$

In second case,

$$W_2 = \frac{1}{2} k(3x^2) - \frac{1}{2} kx^2 = 8 \times \frac{1}{2} kx^2 = 8W_1.$$

71. (3) From the given equation $k = 8 \text{ ms}^{-1}$ and $\omega = 4 \text{ rad/s}$

$$\text{Velocity of wave} = \frac{\omega}{k}$$

$$v = \frac{4}{8} = 0.5 \text{ m/s}$$

72. (1) $R_s = \frac{I_g R_g}{I - I_g}$

$$\frac{1 \times 10^{-3} \times 100}{10 - 0.001} \approx 0.01 \Omega$$

73. (4) Condition for minimum path diff. = $(2n - 1) \frac{\lambda}{2}$.

$$\text{For } n = 3, \text{ path diff.} = (2 \times 3 - 1) \frac{\lambda}{2} = \frac{5\lambda}{2}$$

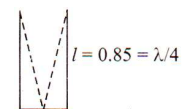
74. (1)

75. (3) Energy released

$$= (80 \times 7 + 120 \times 8) - (200 \times 6.5) = 220 = 2 \times 110 \text{ MeV.}$$

76. (1) In fundamental mode

$$\frac{\lambda}{4} = 0.85$$



$$\lambda = 4 \times 0.85$$

$$f = \frac{v}{\lambda} = \frac{340}{4 \times 0.85} = 100 \text{ Hz}$$

$$\therefore \text{Possible frequencies} = 100 \text{ Hz, } 300 \text{ Hz}$$

77. (2)

78. (2) As the measured length is 3.50 cm, the instrument must have a least count of 0.01 cm or 0.1 mm.

For the vernier callipers, 1 MSD = 1 mm

10 VSD = 9 MSD

$$\Rightarrow 1 \text{ VSD} = \frac{9}{10} \text{ MSD} = \frac{9}{10} \text{ mm}$$

Least count = 1 MSD - 1 VSD

$$= \left(1 - \frac{9}{10}\right) \text{ mm} = \frac{1}{10} \text{ mm} = 0.1 \text{ mm}$$

79. (1) Rocket propulsion is based on the principle of Newton's third law, that is every action has equal and opposite reaction.

Hence, the correct answer is option (1).

80. (2)

81. (2) According to question, 7th coin is under the weight of 5 coins above it.

Hence, reaction of the 7th coin on the 8th coin

= force on the 7th coin due to 8th coin

= weight of 5 coins

$$= 5 \text{ mg} = 5 \times 20 \times 10^{-3} \times 10 = 1 \text{ N}$$

$$[\because m = 20 \text{ g} = 20 \times 10^{-3} \text{ kg}]$$

82. (4) $K.E. = hf - \phi$

$$\tan \theta = h.$$

83. (2) Applying energy conservation between A & B

$$\frac{1}{2} mV_L^2 = \frac{1}{2} mV_H^2 + mg(2L)$$

$$\therefore V_L = \sqrt{5gL}$$

$$\text{so, } V_H = \sqrt{gL}$$

$$\frac{(K.E.)_A}{(K.E.)_B} = \frac{\frac{1}{2} m(\sqrt{5gL})^2}{\frac{1}{2} m(\sqrt{gL})^2} = \frac{5}{1}$$

84. (2) S_1 in first $(p - 1)$ sec

S_2 in first p sec

$$S_1 = \frac{1}{2} a(p - 1)^2$$

$$S_2 = \frac{1}{2} a(p)^2$$

$$S_1 + S_2 = \frac{1}{2} at^2$$

$$(p - 1)^2 + p^2 = t^2$$

$$t = \sqrt{2p^2 + 1 - 2p}$$

85. (3) Here, $N = 2000$ A = 0.2 m².

$$\omega = 400 \text{ s}^{-1}, B = 0.5 \text{ T}, e_0 = ?$$

$$e_0 = NAB\omega = 2000 \times 0.2 \times 0.5 \times 400 = 80000 \text{ V} = 80 \text{ kV}.$$

SECTION - B (Attempt Any 10 Questions)

86. (3) Here, $T = 4 \text{ s}$; Displacement $y = a - a/2 = a/2$

$$\text{As } y = a \cos \omega t = a \cos \frac{2\pi}{T} t$$

$$\therefore \frac{a}{2} = a \cos \frac{2\pi}{4} t \text{ or } \cos \frac{2\pi}{4} = \frac{1}{2} = \cos \frac{\pi}{3}$$

$$\text{or } \frac{2\pi t}{4} = \frac{\pi}{3} \text{ or } t = \frac{2}{3} \text{ s}.$$

87. (1) $n_e = 8 \times 10^{13} / \text{cm}^3$, $n_h = 5 \times 10^{13} / \text{cm}^3$
 $\mu_e = 25000 \text{ cm}^2 \text{ V}^{-1} \text{ sec}^{-1}$, $\mu_h = 1000 \text{ cm}^2 \text{ V}^{-1} \text{ sec}^{-1}$

$$\sigma = n_e \mu_e e + n_h \mu_h e$$

$$= (8 \times 10^{13} \times 25000 + 5 \times 10^{13} \times 1000) \times 1.6 \times 10^{-19} = 328 \times 10^{-3} \text{ mho/cm}$$

$$n_e > n_h (n\text{-type})$$

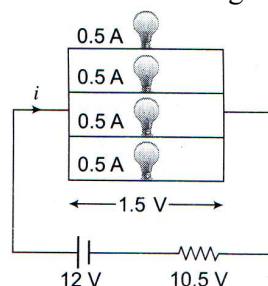
88. (3) Capacitive reactance, $X_C = \frac{1}{\omega C} = \frac{1}{2\pi\nu C}$

$$\Rightarrow X_L \propto \frac{1}{\nu}$$

With increases in frequency, X_C decreases.

Hence, option (3) represents the correct graph.

89. (2) For normal brightness of each bulb see following circuit. current through each bulb = 0.5 A



So main current $i = 2 \text{ A}$

Also, voltage across the combination = 1.5 V

So voltage across the resistance = 10.5 V

Hence, for resistance, $V = iR \Rightarrow 10.5 = 2 \times R = R$

$$= \frac{21}{4} \Omega$$

90. (3) Charge on capacitor plates without the dielectric is $Q = CV = (5 \times 10^{-6} \text{ F}) \times 1\text{V} = 5 \times 10^{-6} \text{ C} = 5 \mu\text{C}$

The capacitance after the dielectric is introduced is

$$C' = \frac{\epsilon_0 A}{d - \left(t - \frac{t}{K}\right)} = \frac{\epsilon_0 A / d}{1 - \left(\frac{t - \frac{t}{K}}{d}\right)}$$

$$= \frac{C}{1 - \left(\frac{t - \frac{t}{K}}{d}\right)} = \frac{5\mu\text{F}}{1 - \left(\frac{4\text{cm} - \frac{4\text{cm}}{4}}{6\text{cm}}\right)} = \frac{5\mu\text{F}}{1 - \left(\frac{4-1}{6}\right)} = 10\mu\text{F}$$

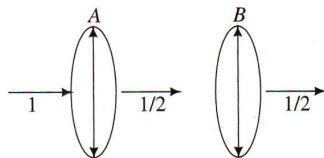
∴ Charge on capacitor plates now will be

$$Q' = C'V = 10\mu\text{F} \times 1\text{V} = 10\mu\text{C}$$

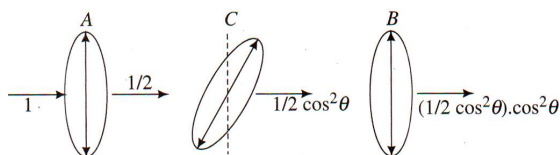
Additional charge transferred =

$$Q' - Q = 10\mu\text{C} - 5\mu\text{C} = 5\mu\text{C}$$

91. (4) Axis of transmission of A and B are parallel.



Now



$$\frac{I}{2} \cos^4 \theta = \frac{I}{8} \Rightarrow \cos^4 \theta = \frac{1}{4}$$

$$\cos \theta = \frac{1}{\sqrt{2}} \Rightarrow \theta = 45^\circ$$

92. (2) $[C^2LR] = \left[C^2 L^2 \frac{R}{L} \right] = \left[(LC)^2 \left(\frac{R}{L} \right) \right]$

and we know that frequency of LC circuits is given by

$f = \frac{1}{2\pi\sqrt{LC}}$ i.e., the dimension of LC is equal to $[T^2]$ and $[L/R]$ gives the time constant of L-R circuit so the dimension of L/R is equal to $[T]$.

By substituting the above dimensions in the given

formula $\left[(LC)^2 \left(\frac{R}{L} \right) \right] = [T^2]^2 [T^{-1}] = [T^3]$.

93. (4) $\frac{1}{f_{air}} = (\mu - 1) \left(\frac{1}{R_1} - \frac{1}{R_2} \right)$

Here, $\frac{1}{f_i} = (1.6 - 1) \left(\frac{1}{\infty} - \frac{1}{20} \right)$

$$f_i = \frac{-100}{3} \text{ So } \frac{1}{f} = \frac{1}{f_1} + \frac{1}{f_2} + \frac{1}{f_3} = \frac{-3}{100} + \frac{-3}{100} + \frac{1}{20}$$

$$\therefore f = -100 \text{ cm.}$$

94. (2) Taking parts A and B as two bodies of same system,

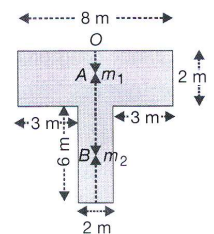
$$m_1 = l \times b \times \sigma = 8 \times 2 \times \sigma = 16\sigma$$

$$m_2 = l \times b \times \sigma = 6 \times 2 \times \sigma = 12\sigma$$

Choosing O as the origin,

$$x_1 = 1\text{m}, x_2 = 2 + 3 = 5\text{m}$$

$$= \frac{16\sigma \times 1 + 12\sigma \times 5}{16\sigma + 12\sigma} = \frac{19}{7}$$



95. (1) For $PV^x = \text{constant}$

If work done by gas is asked then

$$W = \frac{nR\Delta T}{1-x}$$

Here $x = \frac{3}{2}$

$$\therefore W = \frac{P_2V_2 - P_1V_1}{-1/2}$$

$$= 2(P_1V_1 - P_2V_2) \dots \text{Option (1) is correct.}$$

96. (2) $T_c = 260 = 10 \times 10 + 10(10 + a)$

$$\Rightarrow a = 6 \text{ m/s}^2$$

97. (4) \vec{B} is measured in tesla ; \vec{H} is measured in ampere/meter, μ is measured in tesla metre/ampere. χ_m has no units.

98. (3) Surface tension will be less as temperature increases

$$h = \frac{2T \cos \theta}{\rho g r}$$

Height of capillary rise will be smaller in hot water and larger in cold water.

99. (2) $v = \sqrt{\frac{2gh}{1 + \frac{k^2}{R^2}}}$

For a solid sphere,

$$\frac{k^2}{R^2} = \frac{2}{5} = \frac{2}{5}; v = \sqrt{\frac{2 \times 10 \times 7}{\left(1 + \frac{2}{5}\right)}} = 10 \text{ m/s}$$

100. (1) $B = \frac{E}{c} = \frac{19.2}{3 \times 10^8} = 6.4 \times 10^{-8} T$

$$\hat{E} \times \hat{B} = \hat{i} \Rightarrow \hat{B} = \hat{k}$$

BOTANY

101. (4) (11th Para 8.5.1, Page no.131)
102. (4) (NCERT 12th Page no- 25, Last Paragraph, Line no-13,14)
103. (3) (NCERT 12th, Page no- 23, Paragraph-2, line-6,7,8,9)
104. (3) (NCERT XIIth Pg 88, based on Pedigree Analysis)
105. (1) (NCERT XI page no. 218, 4th paragraph 2nd last line)
106. (3) (12th NCERT Page no.249, 3rd para pyramid based)
107. (2) (11th Para 8.5.1, Page no.131)
108. (2) (11th Para 10.1.1 based , Page no.163)
109. (4) (11th Para 10.4.1, Page no.168)
110. (2) (NCERT XI Pg.230, 14.3, 3rd Para, 1st line)
111. (4) [NCERT XI, Page No. 79; Sub-topic 5.9.1]
112. (2) (11th NCERT P.K. concept ,3.2 to 3.3)
113. (3) (11th NCERT Page no 31 to 39 Conceptual)
114. (1) (NCERT XI Pg.236, 14.7, 1st Para, 1st line)
115. (3) (11th NCERT PK, Conceptual, bryo to gymno)
116. (3) [NCERT class XI, Page no. 88, Point 6.2.1- Last paragraph]

117. (4) [NCERT class XI, Page 250, Point 15.4.3.4, (Third paragraph)]
118. (3) (Pg 122, Based on gel electrophoresis)
119. (2) (NCERT XIIth Pg 79, based on Figure 5.7)
120. (2) (NCERT 12th, Page no- 24, Paragraph-2.2.2, Line no- 1, 2, 6,7)
121. (4) (NCERT 11th, Page no- 21, paragraph-2.2.4, Line no- 28-30)
122. (3) (NCERT 11th, Page no- 24, paragraph-2.3.3, Line no- 12)
123. (4) (NCERT XIIth Pg 84, based on Figure 5.11)
124. (1) [NCERT class XI, Page 243, First paragraph]
125. (2) [NCERT class XI, Page 248, point 15.4.3.1 (Last paragraph)]

126. (3) [NCERT XI, Page No. 76; Sub-topic]
127. (4) (NCERT XIIth Pg 83, Para 2)
128. (4) (NCERT 11th, Page no- 7, 2nd Paragraph, Line no- 10-12)
129. (2) (NCERT 11th, Page no- 22, 1st paragraph, Line no- 7,8)
130. (4) [NCERT XI, Newly added families)
131. (1) (NCERT XIIth Pg 111, Based on RNA processing (Splicing))
132. (3) (NCERT XIIth Pg Pg 118, Para 3, Line 1-2)
133. (2) (NCERT XIIth Pg Pg 97, Based on DNA structure)
134. (3) (NCERT XIIth Pg 96, Para 4, Line 11)
135. (4) (NCERT XI page no. 219, 2nd paragraph)

SECTION - B (Attempt Any 10 Questions)

136. (2) (11th Para 8.5.11 Page no.140)
137. (3) (NCERT 11th, Page no- 23,24, Conceptual)
138. (3) (NCERT 12th, Page no22, 1st paragraph, Line no- 1-5 Concept based)
139. (4) (12th NCERT Page no.244-last para to 245-2nd para)
140. (2) (11th NCERT PK Page no.33 Table3.1, page 34 to 39, conceptual.)
141. (1) [NCERT class XI, page no. 89 (First paragraph), 94 (Second paragraph) + NCERT Exemplar, Question no. 03]

142. (1) [NCERT XI, Page No. 79 & 80; Sub-topic 5.9.1 & 5.9.2]
143. (3) (NCERT XIIth Pg 85, based on POLYGENIC INHERITANCE)
144. (2) (11th Para 10.5, Page no.170)
145. (2) (NCERT XI Pg.243, Fig.14.4, Diagram)
146. (1) (11th Para 8.5.6 concept based, Page no.136)
147. (3) (NCERT XIIth Pg 113, Point 6.6.1)
148. (4) (NCERT XI page no. 220, sub-topic 13.9 – 3rd paragraph)
149. (1) (NCERT XIIth Pg 80, based on Law of Independent Assortment)
150. (2) (NCERT 11th, Page no- 9, Paragraph- 1.3.1, Line no- 5,6)

ZOOLOGY

Section - A (35 Questions)

151. (1) (NCERT 11th page no 101, para 1)
152. (2) (NCERT 12th page no 62, para 1)
153. (3) (NCERT 12th, page no 60, para 2)
154. (2) (NCERT 12th, Page no- 137, 2nd paragraph, line no-3-8)
155. (2) (NCERT 11th, Page no- 144, 2nd Paragraph, Line no- 1)
156. (3) (NCERT 11th, Page no- 148, Paragraph- 9.5, Line no- 13,14)
157. (1) (NCERT XI Page No. 47; 4.1.2)
158. (1) (NCERT XI Page No. 59; 1st Paragraph)
159. (3) (NCERT 11th page 118, para 2, line 14)
160. (4) (12th Para 10.1, Page no.186)
161. (2) (NCERT Pg. No.154)
162. (2) (NCERT Pg. No. 284)
163. (3) (NCERT 12th, Page no- 135, Last paragraph, line no- 7,8)
164. (3) (12th NCERT Page no.234 2nd para.)
165. (3) (NCERT 12th HR)
166. (2) (NCERT 12th page no 46, para 3)
167. (2) (12th Para 10.1, Page no.186)
168. (2) [NCERT P.No.214, Last para]
169. (3) [NCERT Exemplar,]

170. (4) (NCERT Pg. No. 153-154)
171. (4) (NCERT Pg. No.281)
172. (1) [NCERT P.No.306, 1st Para]
173. (2) [NCERT P.No.316, Human Neural System, 15th Line]
174. (2) [NCERT P.No.320 CNS 2nd para]
175. (2) [NCERT P.No.195 and 202,]
176. (3) (NCERT page no. 262 (ii) sp.-Area relation, summary 268, 3rd para.)
177. (1) (NCERT XI Page No. 291, 1st paragraph)
178. (1) (NCERT XI Page No. 292, 1st line)
179. (3) (NCERT XI Page No. 333, 9th line of 2nd paragraph)
180. (1) (NCERT XI Page No. 333, 15th line of 2nd paragraph)
181. (3) [NCERT P.No.310 12th Line]
182. (2) [NCERT P.No.312, Synovial Joints]
183. (1) (NCERT Pg. No. 284-285)
184. (2) (NCERT Pg. No. 161)
185. (1) (NCERT 12th, Page no- 127, 3rd paragraph, line no- 8,9)
(NCERT 12th, Page no- 130, Last paragraph, line no- 9,10)

SECTION - B (Attempt Any 10 Questions)

186. (4) (NCERT Pg. No. 148)
187. (2) (NCERT 11th page 115, para 1)
188. (3) [NCERT P.No.302, 2nd para]
189. (2) [NCERT P.No.312]
190. (3) [NCERT P.No.209, Pest Resistant Plants 1st para]
191. (2) (12th NCERT Page no.265, 15.2.1, 2nd para.)
192. (4) (12th NCERT Page no.263, 3rd para)
193. (2) (NCERT 12th, p.no.52,53)
194. (3) (NCERT 12th page 64, 4.5 infertility)
195. (1) (NCERT Pg. No. 158-159)
196. (2) (NCERT 12th, Page no- 137, 3rd paragraph, line no-1 and 2)
197. (3) (NCERT XI NCERT Conceptual)
198. (3) (NCERT 11th, Page no- 159, Paragraph- 9.12.6, Line no- 16, 17)
199. (1) (NCERT Pg. No. 272)
200. (4) (12th Para 10.2.3, Page no. 183)